

Polyset RTA-1- 24oz ICP Building Solutions Group

Version No: 2.7.2.1

Safety Data Sheet according to OSHA HazCom Standard (2012) requirements

Issue Date: **04/28/2021**Print Date: **04/28/2021**S.GHS.USA.EN

SECTION 1 Identification

Р	'ro	d	u	ct	: 1	d	eı	nti	ifi	er	

Product name	Polyset RTA-1- 24oz		
Synonyms	SDS A16163		
Proper shipping name	Aerosols, non-flammable, (each not exceeding 1 L capacity)		
Other means of identification	Not Available		

Recommended use of the chemical and restrictions on use

Relevant identified uses	One Component Polyurethane Foam Adhesive
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Name, address, and telephone number of the chemical manufacturer, importer, or other responsible party

Registered company name	ICP Building Solutions Group		
Address	2775 Barber Road Norton Ohio 44203 United States		
Telephone	+1 330 753 4585 +1 800 321 5585		
Fax	Not Available		
Website http://www.icpgroup.com/			
Email sds@icpgroup.com			

Emergency phone number

<u> </u>	<u> </u>		
Association / Organisation	CHEMTEL		
Emergency telephone numbers	+1 800 255 3924		
Other emergency telephone numbers	+1 813 248 0585		

SECTION 2 Hazard(s) identification

Classification of the substance or mixture

NFPA 704 diamond



Note: The hazard category numbers found in GHS classification in section 2 of this SDSs are NOT to be used to fill in the NFPA 704 diamond. Blue = Health Red = Fire Yellow = Reactivity White = Special (Oxidizer or water reactive substances)

Classification

Eye Irritation Category 2A, Respiratory Sensitizer Category 1, Specific target organ toxicity - repeated exposure Category 2, Acute Toxicity (Inhalation) Category 4, Specific target organ toxicity - single exposure Category 3 (respiratory tract irritation), Skin Corrosion/Irritation Category 2, Gas under Pressure (Compressed gas), Skin Sensitizer Category 1

Label elements

Hazard pictogram(s)







Signal word

ord Dange

Hazard statement(s)

H319 Causes serious eye irritation.				
H334 May cause allergy or asthma symptoms or breathing difficulties if inhaled.				

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H373	May cause damage to organs through prolonged or repeated exposure. (Respiratory system) (Inhalation)		
H332	Harmful if inhaled.		
H335	May cause respiratory irritation.		
H315	H315 Causes skin irritation.		
H280	H280 Contains gas under pressure; may explode if heated.		
H317	May cause an allergic skin reaction.		

Hazard(s) not otherwise classified

Not Applicable

Precautionary statement(s) Prevention

P102	Meep Out of Reach of Children.		
P202	Do not handle until all safety precautions have been read and understood.		
P251	Pressurized Container: Do not pierce or burn, even after use.		
P260	Do not breathe dust/fume/gas/mist/vapors/spray		
P262	Do not get in eyes, on skin, or on clothing.		
P271	P271 Use only outdoors or in a well-ventilated area.		
P273	P273 Avoid release to the environment		
P280	Wear protective gloves, protective clothing and eye protection.		

Precautionary statement(s) Response

P302+P352+P333+P313	IF ON SKIN: Wash with plenty of soap and water. If skin irritation or rash occurs: Get medical attention		
P304+P340	P304+P340 IF INHALED: Remove victim to fresh air and keep at rest in a position comfortable for breathing.		
P308+P313 IF exposed or concerned: Get medical advice/attention.			
P305+P351+P338 IF IN EYES: Rinse cautiously with water for several minutes. Remove contact lenses, if present and easy to do. Continue rinsing.			
P342+P311 If experiencing respiratory symptoms: Call a POISON CENTER or doctor/physician.			

Precautionary statement(s) Storage

P405	Store locked up.	
P410+P403	Protect from sunlight. Store in a well-ventilated place.	
P412 Do not expose to temperatures exceeding 50°C/122°F.		

Precautionary statement(s) Disposal

P501	Dispose of contents/container in accordance with applicable local/regional/national/international regulations.

SECTION 3 Composition / information on ingredients

Substances

See section below for composition of Mixtures

Mixtures

CAS No	%[weight]	Name
25322-69-4	15-30	polypropylene glycol
13674-84-5*	10-15	tris(2-chloroisopropyl)phosphate
29118-24-9	10-30	1,3,3,3-tetrafluoropropene
101-68-8	10-30	4.4'-diphenylmethane diisocyanate (MDI)
9016-87-9	10-30	MDI oligomer

The specific chemical identity and/or exact percentage (concentration) of composition has been withheld as a trade secret.

SECTION 4 First-aid measures

Description of first aid measures

If aerosols come in contact with the eyes:

Eye Contact

Immediately hold the eyelids apart and flush the eye continuously for at least 15 minutes with fresh running water.

- Figure complete irrigation of the eye by keeping eyelids apart and away from eye and moving the eyelids by occasionally lifting the upper and lower lids.
- Transport to hospital or doctor without delay.
- ▶ Removal of contact lenses after an eye injury should only be undertaken by skilled personnel.

Skin Contact

If solids or aerosol mists are deposited upon the skin:

- Flush skin and hair with running water (and soap if available).
- Remove any adhering solids with industrial skin cleansing cream.
- DO NOT use solvents
- ▶ Seek medical attention in the event of irritation.

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Following uptake by inhalation, move person to an area free from risk of further exposure. Oxygen or artificial respiration should be administered as needed. Asthmatic-type symptoms may develop and may be immediate or delayed up to several hours. Treatment is essentially symptomatic. A physician should be consulted.

If aerosols, fumes or combustion products are inhaled:

Prosthese is uch as false teeth, which may block airway, should be removed, where possible, prior to initiating first aid procedures.

If breathing is shallow or has stopped, ensure clear airway and apply resuscitation, preferably with a demand valve resuscitator, bag-valve mask device, or pocket mask as trained. Perform CPR if necessary.

Transport to hospital, or doctor.

Not considered a normal route of entry.

Avoid giving milk or oils.

Avoid giving alcohol.

Most important symptoms and effects, both acute and delayed

See Section 11

Indication of any immediate medical attention and special treatment needed

for intoxication due to Freons/ Halons;

- A: Emergency and Supportive Measures
- Maintain an open airway and assist ventilation if necessary
- Treat coma and arrhythmias if they occur. Avoid (adrenaline) epinephrine or other sympathomimetic amines that may precipitate ventricular arrhythmias. Tachyarrhythmias caused by increased myocardial sensitisation may be treated with propranolol, 1-2 mg IV or esmolol 25-100 microgm/kg/min IV.
- ► Monitor the ECG for 4-6 hours
- B: Specific drugs and antidotes:
- ► There is no specific antidote
- C: Decontamination
- Inhalation; remove victim from exposure, and give supplemental oxygen if available.
- Ingestion; (a) Prehospital: Administer activated charcoal, if available. DO NOT induce vomiting because of rapid absorption and the risk of abrupt onset CNS depression. (b) Hospital: Administer activated charcoal, although the efficacy of charcoal is unknown. Perform gastric lavage only if the ingestion was very large and recent (less than 30 minutes)
- D: Enhanced elimination:

▶ There is no documented efficacy for diuresis, haemodialysis, haemoperfusion, or repeat-dose charcoal POISONING and DRUG OVERDOSE, Californian Poison Control System Ed. Kent R Olson; 3rd Edition

- ▶ Do not administer sympathomimetic drugs unless absolutely necessary as material may increase myocardial irritability.
- No specific antidote
- Because rapid absorption may occur through lungs if aspirated and cause systematic effects, the decision of whether to induce vomiting or not should be made by an attending physician.
- If lavage is performed, suggest endotracheal and/or esophageal control.
- ▶ Danger from lung aspiration must be weighed against toxicity when considering emptying the stomach.
- Treatment based on judgment of the physician in response to reactions of the patient
- Polyethylene glycols are generally poorly absorbed orally and are mostly unchanged by the kidney.
- Dermal absorption can occur across damaged skin (e.g. through burns) leading to increased osmolality, anion gap metabolic acidosis, elevated calcium, low ionised calcium, CNS depression and renal failure.
- Treatment consists of supportive care.

[Ellenhorn and Barceloux: Medical Toxicology]

Treat symptomatically.

For sub-chronic and chronic exposures to isocyanates:

- This material may be a potent pulmonary sensitiser which causes bronchospasm even in patients without prior airway hyperreactivity.
- ▶ Clinical symptoms of exposure involve mucosal irritation of respiratory and gastrointestinal tracts
- Conjunctival irritation, skin inflammation (erythema, pain vesiculation) and gastrointestinal disturbances occur soon after exposure.
- Pulmonary symptoms include cough, burning, substernal pain and dyspnoea
- ► Some cross-sensitivity occurs between different isocyanates.
- Noncardiogenic pulmonary oedema and bronchospasm are the most serious consequences of exposure. Markedly symptomatic patients should receive oxygen, ventilatory support and an intravenous line.
- ▶ Treatment for asthma includes inhaled sympathomimetics (epinephrine [adrenalin], terbutaline) and steroids.
- Activated charcoal (1 g/kg) and a cathartic (sorbitol, magnesium citrate) may be useful for ingestion.
- ▶ Mydriatics, systemic analgesics and topical antibiotics (Sulamyd) may be used for corneal abrasions.
- There is no effective therapy for sensitised workers.

[Ellenhorn and Barceloux; Medical Toxicology]

NOTE: Isocyanates cause airway restriction in naive individuals with the degree of response dependant on the concentration and duration of exposure. They induce smooth muscle contraction which leads to bronchoconstrictive episodes. Acute changes in lung function, such as decreased FEV1, may not represent sensitivity. [Karol & Jin, Frontiers in Molecular Toxicology, pp 56-61, 1992]

Personnel who work with isocyanates, isocyanate prepolymers or polyisocyanates should have a pre-placement medical examination and periodic examinations thereafter, including a pulmonary function test. Anyone with a medical history of chronic respiratory disease, asthmatic or bronchial attacks, indications of allergic responses, recurrent eczema or sensitisation conditions of the skin should not handle or work with isocyanates. Anyone who develops chronic respiratory distress when working with isocyanates should be removed from exposure and examined by a physician. Further exposure must be avoided if a sensitivity to isocyanates or polyisocyanates has developed.

SECTION 5 Fire-fighting measures

Extinguishing media

- Small quantities of water in contact with hot liquid may react violently with generation of a large volume of rapidly expanding hot sticky semi-solid foam.
- Presents additional hazard when fire fighting in a confined space.
- Cooling with flooding quantities of water reduces this risk.
- Dry chemical powder.
- BCF (where regulations permit).
- ► Carbon dioxide.

SMALL FIRE:

▶ Water spray, dry chemical or CO2

LARGE FIRE:

▶ Water spray or fog.

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Fire Incompatibility Avoid contamination with oxidising agents i.e. nitrates, oxidising acids, chlorine bleaches, pool chlorine etc. as ignition may result Special protective equipment and precautions for fire-fighters GENERAL Fire Fighting ▶ Alert Fire Brigade and tell them location and nature of hazard. Wear full body protective clothing with breathing apparatus. Fight fire from a safe distance, with adequate cover. -Combustible. -Moderate fire hazard when exposed to heat or flame. -When heated to high temperatures decomposes rapidly generating vapour which pressures and may then rupture containers with release of flammable and highly toxic isocyanate vapour. ▶ Containers may explode when heated - Ruptured cylinders may rocket May burn but does not ignite easily. Fire exposed cylinders may vent contents through pressure relief devices thereby increasing vapour concentration.. Fire may produce irritating, poisonous or corrosive gases. Decomposition may produce toxic fumes of: Fire/Explosion Hazard carbon monoxide (CO) carbon dioxide (CO2) isocyanates hydrogen cyanide and minor amounts of nitrogen oxides (NOx) hydrogen fluoride other pyrolysis products typical of burning organic material. WARNING: Aerosol containers may present pressure related hazards.

SECTION 6 Accidental release measures

Personal precautions, protective equipment and emergency procedures

See section 8

Environmental precautions

See section 12

Methods and material for containment and cleaning un

Minor Spills	Clean up all spills immediately. Avoid breathing vapours and contact with skin and eyes.
	▶ Wear protective clothing, impervious gloves and safety glasses.
	For isocyanate spills of less than 40 litres (2 m2):
	Evacuate area from everybody not dealing with the emergency, keep them upwind and prevent further access, remove ignition sources and if inside building, ventilate area as well as possible.
	▶ Notify supervision and others as necessary.
	 Put on personal protective equipment (suitable respiratory protection, face and eye protection, protective suit, gloves and impermeable boots).
	Avoid contamination with water, alkalies and detergent solutions.
	Material reacts with water and generates gas, pressurises containers with even drum rupture resulting.
	DO NOT reseal container if contamination is suspected.
Major Spills	Clear area of all unprotected personnel and move upwind.
	Alert Emergency Authority and advise them of the location and nature of hazard.
	Wear full body clothing with breathing apparatus.
	▶ Remove leaking cylinders to a safe place.
	Fit vent pipes. Release pressure under safe, controlled conditions
	▶ Burn issuing gas at vent pipes.
	DO NOT exert excessive pressure on valve; DO NOT attempt to operate damaged valve.
	Clear area of personnel and move upwind.
	▶ Alert Fire Brigade and tell them location and nature of hazard.
	May be violently or explosively reactive.

Personal Protective Equipment advice is contained in Section 8 of the SDS.

SECTION 7 Handling and storage

Precautions for safe handling	
Safe handling	 Avoid all personal contact, including inhalation. Wear protective clothing when risk of exposure occurs. Use in a well-ventilated area.
Other information	Consider storage under inert gas. Cylinders should be stored in a purpose-built compound with good ventilation, preferably in the open. Such compounds should be sited and built in accordance with statutory requirements. The storage compound should be kept clear and access restricted to authorised personnel only.

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Suitable container

Storage incompatibility

- Aerosol dispenser.
- Check that containers are clearly labelled.

• Glycols and their ethers undergo violent decomposition in contact with 70% perchloric acid. This seems likely to involve formation of the glycol perchlorate esters (after scission of ethers) which are explosive, those of ethylene glycol and 3-chloro-1,2-propanediol being more powerful than glyceryl nitrate, and the former so sensitive that it explodes on addition of water.

As a general rule, hydrofluorocarbons tend to be flammable unless they contain more fluorine atoms than hydrogen atoms.

- Avoid magnesium, aluminium and their alloys, brass and steel.
- Avoid reaction with water, alcohols and detergent solutions. Isocyanates are electrophiles, and as such they are reactive toward a variety of nucleophiles including alcohols, amines, and even water. Upon treatment with an alcohol, an isocyanate forms a urethane linkage.
 - A range of exothermic decomposition energies for isocyanates is given as 20-30 kJ/mol.
- The relationship between energy of decomposition and processing hazards has been the subject of discussion; it is suggested that values of energy released per unit of mass, rather than on a molar basis (J/g) be used in the assessment.
- For example, in "open vessel processes" (with man-hole size openings, in an industrial setting), substances with exothermic decomposition energies below 500 J/g are unlikely to present a danger, whilst those in "closed vessel processes" (opening is a safety valve or bursting disk) present some danger where the decomposition energy exceeds 150 J/g.
- Compressed gases may contain a large amount of kinetic energy over and above that potentially available from the energy of reaction produced by the gas in chemical reaction with other substances

SECTION 8 Exposure controls / personal protection

Control parameters

Occupational Exposure Limits (OEL)

INGREDIENT DATA

Source	Ingredient	Material name	TWA	STEL	Peak	Notes
US OSHA Permissible Exposure Limits (PELs) Table Z-1	4,4'-diphenylmethane diisocyanate (MDI)	Methylene bisphenyl isocyanate (MDI)	Not Available	Not Available	0.02 ppm / 0.2 mg/m3	Not Available
US NIOSH Recommended Exposure Limits (RELs)	4,4'-diphenylmethane diisocyanate (MDI)	Methylene bisphenyl isocyanate	0.005 ppm / 0.05 mg/m3	Not Available	0.020 (10-minute) ppm / 0.2 (10-minute) mg/m3	Not Available
US ACGIH Threshold Limit Values (TLV)	4,4'-diphenylmethane diisocyanate (MDI)	Methylene bisphenyl isocyanate	0.005 ppm	Not Available	Not Available	Not Available

Emergency Limits

Ingredient	TEEL-1	TEEL-2	TEEL-3
polypropylene glycol	30 mg/m3	330 mg/m3	2,000 mg/m3
1,3,3,3-tetrafluoropropene	1,400 ppm	Not Available	Not Available
4,4'-diphenylmethane diisocyanate (MDI)	0.45 mg/m3	Not Available	Not Available
4,4'-diphenylmethane diisocyanate (MDI)	29 mg/m3	40 mg/m3	240 mg/m3
MDI oligomer	0.15 mg/m3	3.6 mg/m3	22 mg/m3

Ingredient	Original IDLH	Revised IDLH
polypropylene glycol	Not Available	Not Available
tris(2-chloroisopropyl)phosphate	Not Available	Not Available
1,3,3,3-tetrafluoropropene	Not Available	Not Available
4,4'-diphenylmethane diisocyanate (MDI)	75 mg/m3	Not Available
MDI oligomer	Not Available	Not Available

Occupational Exposure Banding

Ingredient	Occupational Exposure Band Rating Occupational Exposure Band Limit		
tris(2-chloroisopropyl)phosphate	E	≤ 0.1 ppm	
MDI oligomer	E ≤ 0.1 ppm		
Notes:	Occupational exposure banding is a process of assigning chemicals into specific categories or bands based on a chemical's potency and the		

Exposure controls

Appropriate engineering controls

Engineering controls are used to remove a hazard or place a barrier between the worker and the hazard. Well-designed engineering controls can be highly effective in protecting workers and will typically be independent of worker interactions to provide this high level of protection. The basic types of engineering controls are:

adverse health outcomes associated with exposure. The output of this process is an occupational exposure band (OEB), which corresponds to a

Process controls which involve changing the way a job activity or process is done to reduce the risk.

Personal protection







range of exposure concentrations that are expected to protect worker health.



Eye and face protection

- Safety glasses with side shields.
- Chemical goggles
- Contact lenses may pose a special hazard; soft contact lenses may absorb and concentrate irritants.

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Skin protection See Hand protection below NOTE: The material may produce skin sensitisation in predisposed individuals. Care must be taken, when removing gloves and other protective equipment, to avoid all possible skin contact. ► Contaminated leather items, such as shoes, belts and watch-bands should be removed and destroyed. ▶ Isocyanate resistant materials include Teflon, Viton, nitrile rubber and some PVA gloves. ▶ Protective gloves and overalls should be worn as specified in the appropriate national standard. Contaminated garments should be removed promptly and should not be re-used until they have been decontaminated. Hands/feet protection No special equipment needed when handling small quantities. ► OTHERWISE: ► For potentially moderate exposures: ▶ Wear general protective gloves, eg. light weight rubber gloves. ► For potentially heavy exposures: ▶ Wear chemical protective gloves, eg. PVC. **Body protection** See Other protection below No special equipment needed when handling small quantities. OTHERWISE: Other protection Overalls. ► Skin cleansing cream.

SECTION 9 Physical and chemical properties

Information on basic physical and chemical properties

Appearance			
	Viscous liquid which forms off-white to yellowish foam upon release.		
Physical state	Compressed Gas	Relative density (Water= 1)	1.1
Odour	Slight fluorocarbon odor during curing stage	Partition coefficient n-octanol / water	Not Available
Odour threshold	Not Available	Auto-ignition temperature (°C)	Not Available
pH (as supplied)	Not Available	Decomposition temperature	Not Available
Melting point / freezing point (°C)	Not Available	Viscosity (cSt)	Not Available
Initial boiling point and boiling range (°C)	HFO-1234ze -19°C (-2.2 °F). Other components boil at 93.3°C (200°F)	Molecular weight (g/mol)	Not Available
Flash point (°C)	Not Available	Taste	Not Available
Evaporation rate	Not Available	Explosive properties	Not Available
Flammability	Use of non-flammable propellant	Oxidising properties	Not Available
Upper Explosive Limit (%)	Not Available	Surface Tension (dyn/cm or mN/m)	Not Available
Lower Explosive Limit (%)	Not Available	Volatile Component (%vol)	Not Available
Vapour pressure (kPa)	345	Gas group	Not Available
Solubility in water	Not Applicable	pH as a solution (1%)	Not Available
Vapour density (Air = 1)	Not Available	VOC g/L	0 Minus Exempt Compounds

SECTION 10 Stability and reactivity

Reactivity	See section 7
Chemical stability	 Elevated temperatures. Presence of open flame. Product is considered stable. Presence of elevated temperatures.
Possibility of hazardous reactions	See section 7
Conditions to avoid	See section 7
Incompatible materials	See section 7
Hazardous decomposition products	See section 5

SECTION 11 Toxicological information

Information on toxicological effects

Inhaled

The material can cause respiratory irritation in some persons. The body's response to such irritation can cause further lung damage. Exposure to fluorocarbons can produce non-specific flu-like symptoms such as chills, fever, weakness, muscle pain, headache, chest discomfort, sore throat and dry cough with rapid recovery. High concentrations can cause irregular heartbeats and a stepwise reduction in lung capacity. Inhalation of the vapour is hazardous and may even be fatal

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The use of a quantity of material in an unventilated or confined space may result in increased exposure and an irritating atmosphere developing. Before starting consider control of exposure by mechanical ventilation. Inhalation of toxic gases may cause: ▶ Central Nervous System effects including depression, headache, confusion, dizziness, stupor, coma and seizures; respiratory: acute lung swellings, shortness of breath, wheezing, rapid breathing, other symptoms and respiratory arrest; heart: collapse, irregular heartheats and cardiac arrest: gastrointestinal: irritation, ulcers, nausea and vomiting (may be bloody), and abdominal pain. The vapour/mist may be highly irritating to the upper respiratory tract and lungs; the response may be severe enough to produce bronchitis and pulmonary oedema. Possible neurological symptoms arising from isocyanate exposure include headache, insomnia, euphoria, ataxia, anxiety neurosis, depression and paranoia. Gastrointestinal disturbances are characterised by nausea and vomiting. WARNING: Intentional misuse by concentrating/inhaling contents may be lethal. There is strong evidence to suggest that this material can cause, if inhaled once, very serious, irreversible damage of organs, Inhalation of aerosols (mists, fumes), generated by the material during the course of normal handling, may produce severely toxic effects. Relatively small amounts absorbed from the lungs may prove fatal. Adverse effects associated with the administration of central nervous system stimulants include shortness of breath, coughing, spasm of the bronchi and spasm of the throat (larynx). Muscular involvement may produce symptoms ranging from twitching to spasticity or seizures. Headache, dizziness and confusion may also result, as can a very high fever or a sensation of warmth. If swallowed, the toxic effects of glycols (dihydric alcohols) are similar to those of alcohol, with depression of the central nervous system, nausea, Ingestion vomiting, and degenerative changes in the liver and kidney. Overexposure to non-ring alcohols causes nervous system symptoms. These include headache, muscle weakness and inco-ordination, giddiness, confusion, delirium and coma. Not normally a hazard due to physical form of product. Considered an unlikely route of entry in commercial/industrial environments The material may accentuate any pre-existing dermatitis condition Skin contact with the material may damage the health of the individual; systemic effects may result following absorption. Spray mist may produce discomfort Fluorocarbons remove natural oils from the skin, causing irritation, dryness and sensitivity. Skin Contact Open cuts, abraded or irritated skin should not be exposed to this material Entry into the blood-stream, through, for example, cuts, abrasions or lesions, may produce systemic injury with harmful effects. Examine the skin prior to the use of the material and ensure that any external damage is suitably protected. There is some evidence to suggest that the material may cause moderate inflammation of the skin either following direct contact or after a delay of some time. Repeated exposure can cause contact dermatitis which is characterised by redness, swelling and blistering. Not considered to be a risk because of the extreme volatility of the gas. This material may produce eye irritation in some persons and produce eye damage 24 hours or more after instillation. Moderate inflammation Eye may be expected with redness; conjunctivitis may occur with prolonged exposure. Long-term exposure to respiratory irritants may result in airways disease, involving difficulty breathing and related whole-body problems. Inhaling this product is more likely to cause a sensitisation reaction in some persons compared to the general population. Skin contact with the material is more likely to cause a sensitisation reaction in some persons compared to the general population. Toxic: danger of serious damage to health by prolonged exposure through inhalation, in contact with skin and if swallowed. This material can cause serious damage if one is exposed to it for long periods. It can be assumed that it contains a substance which can produce severe defects. Substance accumulation, in the human body, may occur and may cause some concern following repeated or long-term occupational exposure. This product contains a polymer with a functional group considered to be of high concern. Isothiocyanates may cause hypersensitivity of the skin and airways. The reactivity of an epoxide intermediate may be the reason for the cancer-causing properties of halogenated oxiranes. It is reported that 1,1-dichloroethyne, vinyl chloride, trichloroethylene, tetrachloroethylene and chloroprene all cause cancer. Chronic Generally speaking, substances with one halogen substitution show higher potential to cause cancer compared to substances with two. Main route of exposure to the gas in the workplace is by inhalation. Persons with a history of asthma or other respiratory problems or are known to be sensitised, should not be engaged in any work involving the handling of isocyanates The chemistry of reaction of isocyanates, as evidenced by MDI, in biological milieu is such that in the event of a true exposure of small MDI doses to the mouth, reactions will commence at once with biological macromolecules in the buccal region and will continue along the digestive tract prior to reaching the stomach. Reaction products will be a variety of polyureas and macromolecular conjugates with for example mucus,

proteins and cell components.

Animal testing shows that polymeric MDI can damage the nasal cavities and lungs, causing inflammation.and increased cell growth. There has been concern that this material can cause cancer or mutations, but there is not enough data to make an assessment.

Fluorocarbons can cause an increased risk of cancer, spontaneous abortion and birth defects.

	TOVIOLTY	IDDITATION
Polyset RTA-1- 24oz	TOXICITY	IRRITATION
,	Not Available	Not Available
	TOVICITY	IDDITATION
	TOXICITY	IRRITATION
	Dermal (rabbit) LD50: >3000 mg/kg ^[1]	Eye: no adverse effect observed (not irritating) ^[1]
polypropylene glycol	Inhalation(Rat) LC50; >2.34 mg/l4h ^[1]	Skin (rabbit): 500 mg mild
	Oral(Rat) LD50; >2000 mg/kg ^[1]	Skin: no adverse effect observed (not irritating) ^[1]
	TOXICITY	IRRITATION
tris(2-	Dermal (rabbit) LD50: >2000 mg/kg ^[1]	Eye (rabbit): non-irritating*
chloroisopropyl)phosphate	Inhalation(Rat) LC50; >4.6 mg/l4h[2]	Skin (rabbit): mild (24 h):
	Oral(Rat) LD50; >500 mg/kg ^[1]	
	TOXICITY	IRRITATION
1,3,3,3-tetrafluoropropene	Inhalation(Rat) LC50; >1157.752 ppm4h ^[2]	Not Available

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	TOXICITY	IRRITATION	
	Dermal (rabbit) LD50: >6200 mg/kg ^[2]	Dermal Sensitise	er *
4,4'-diphenylmethane	Inhalation(Rat) LC50; 0.368 mg/L4h ^[1]		effect observed (not irritating) ^[1]
diisocyanate (MDI)	Oral(Rat) LD50; >2000 mg/kg ^[1]	Skin (rabbit): 500	
		` ,	fect observed (irritating) ^[1]
		, ,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,	
	TOXICITY	IRRITATION	
MDI oligomer	Dermal (rabbit) LD50: >9400 mg/kg ^[2]	Eye (rabbit): 100	mg - mild
	Inhalation(Rat) LC50; 0.49 mg/L4h ^[2]		
	Oral(Rat) LD50; 43000 mg/kg ^[2]		
Legend:	Value obtained from Europe ECHA Registered Sub specified data extracted from RTECS - Register of To:	•	ined from manufacturer's SDS. Unless otherwise
POLYPROPYLENE GLYC		xidised surfactant is non-sensitizing, relations of the contact causing inflammation.	nany of the oxidation products are sensitisers. The Repeated or prolonged exposure to irritants may
TRIS HLOROISOPROPYL)PHOSPHA	Non-chlorinated triphosphates have varying chem a source of potential exposure (human and enviro ingredient in rubber or plastic to the outer surface For tris(2-chloro-1-methylethyl)phosphate (TCPP The flame retardant product supplied in the EU, n isomers in this reaction mixture are not separated Alkyl esters of phosphoric acid exhibit a low to me	nical, physical, toxicological and environmental) to triphosphate plasticisers after curing.) narketed as TCPP, is actually a reaction marketed. The individual componentates acute toxicity and metabolisee	onmental properties. Blooming has been identified as / flame retardants. Blooming is the movement of an on mixture containing four isomers. The individual ents are never produced as such. d. From studies done on mice, they are not likely to n effect on newborn rats at high doses to the pregnant.
1,3,3,3-TETRAFLUOROPROPE	induced serious toxic effects. Rats and rabbits did to high levels of HFO-1234ze. Based on a series sensitisation was observed in dogs with exposure	exposures at levels higher than 10% he d not exhibit any serious toxic, develor of mutagenicity and genomics studies as up to 120,000 ppm; repeated dose aclude negative Ames Test and negati	ave not induced cardiac sensitization to adrenalin no omental or reproductive effects even with exposures i, the cancer risk for HFO-1234ze is low, no cardiac toxicity in rats (13-wk) found mild effects on the hear we human lymphocyte chromosome aberration test;
4,4'-DIPHENYLMETHA DIISOCYANATE (M	Inhalation (human) TCLo: 0.13 ppm/30 mins Eye (rabbit): 0.10 mg moderate		
MDI OLIGOM			
Polyset RTA-1- 24oz 4,4'-DIPHENYLMETHAI DIISOCYANATE (MDI) & N	Asthma-like symptoms may continue for months or even years after exposure to the material ends. This may be due to a non-allergic condition known as reactive airways dysfunction syndrome (RADS) which can occur after exposure to high levels of highly irritating compound. Main criteria for diagnosing RADS include the absence of previous airways disease in a non-atopic individual, with sudden of persistent asthma-like symptoms within minutes to hours of a documented exposure to the irritant. Allergic reactions involving the respiratory tract are usually due to interactions between IgE antibodies and allergens and occur rapidly. Allergic potential of the allergen and period of exposure often determine the severity of symptoms. Some people may be genetically mor prone than others, and exposure to other irritants may aggravate symptoms. Attention should be paid to atopic diathesis, characterised by increased susceptibility to nasal inflammation, asthma and eczema. Exogenous allergic alveolitis is induced essentially by allergen specific immune-complexes of the IgG type; cell-mediated reactions (T lymphocytes) may be involved. Such allergy is of the delayed type with onset up to four hours following exposure. The following information refers to contact allergens as a group and may not be specific to this product. Contact allergies quickly manifest themselves as contact eczema, more rarely as urticaria or Quincke's oedema. The pathogenesis of contact eczema involves a cell-mediated (T lymphocytes) immune reaction of the delayed type. Other allergic skin reactions, e.g. contact		
OLIGOM	The following information refers to contact allerge Contact allergies quickly manifest themselves as	the delayed type with onset up to four ens as a group and may not be specific contact eczema, more rarely as urtica hocytes) immune reaction of the delay	hours following exposure. c to this product. Iria or Quincke's oedema. The pathogenesis of
	The following information refers to contact allerge Contact allergies quickly manifest themselves as contact eczema involves a cell-mediated (T lympl urticaria, involve antibody-mediated immune reac Inhalation of perfluoroalkenes can cause lung injuressure and the production of blood cells. The less toxic.	the delayed type with onset up to four sas as a group and may not be specific contact eczema, more rarely as urtice hocytes) immune reaction of the delay titions. ury, kidney damage, brain changes an potential for causing cancer is the suldisinfectants such as chlorine, chlora some DBPs cause cancer. To date, see	r hours following exposure. c to this product. uria or Quincke's oedema. The pathogenesis of red type. Other allergic skin reactions, e.g. contact d death. Repeated exposures may alter blood oject of speculation. Fluoroalkanes, in contrast, are mines and ozone react with organic and inorganic veral hundred DBPs have been identified.
OLIGOM	The following information refers to contact allergee Contact allergies quickly manifest themselves as contact eczema involves a cell-mediated (T lympl urticaria, involve antibody-mediated immune reac Inhalation of perfluoroalkenes can cause lung injuressure and the production of blood cells. The less toxic. Disinfection byproducts (DBPs) are formed when matter in water. Animal studies have shown that shumerous haloalkanes and haloalkenes have be lsocyanate vapours are irritating to the airways are consciousness and fluid in the lungs. Nervous syinco-ordination, anxiety, depression and paranoia The material may produce moderate eye irritation conjunctivitis. Aromatic and aliphatic diisocyanates may cause and contact and contact and aliphatic diisocyanates may cause and contact allergee contact allergee contact allergee as a contact allergee and contact allergee as contact allergee as contact allergee and contact allergee and contact allergee as contac	the delayed type with onset up to four the delayed type with onset up to four sa sa a group and may not be specific contact eczema, more rarely as urtice hocytes) immune reaction of the delay titions. In the delayed the d	r hours following exposure. c to this product. uria or Quincke's oedema. The pathogenesis of gred type. Other allergic skin reactions, e.g. contact and death. Repeated exposures may alter blood object of speculation. Fluoroalkanes, in contrast, are mines and ozone react with organic and inorganic veral hundred DBPs have been identified. ation-causing activities. wheezing, gasping, severe distress, even loss of e headache, sleep disturbance, euphoria,
Polyset RTA-1- 24oz & 1,3,3 TETRAFLUOROPROPE 4,4'-DIPHENYLMETHAL DIISOCYANATE (MDI) & N	The following information refers to contact allerge Contact allergies quickly manifest themselves as contact eczema involves a cell-mediated (T lympl urticaria, involve antibody-mediated immune reac Inhalation of perfluoroalkenes can cause lung injupressure and the production of blood cells. The less toxic. Disinfection byproducts (DBPs) are formed when matter in water. Animal studies have shown that so Numerous haloalkenes and haloalkenes have be lsocyanate vapours are irritating to the airways and consciousness and fluid in the lungs. Nervous sysinco-ordination, anxiety, depression and paranoia The material may produce moderate eye irritation conjunctivitis. Aromatic and aliphatic diisocyanates may cause a effect. Of the several members of diisocyanates to others produced a harmless outcome. The substance is classified by IARC as Group 3: NOT classifiable as to its carcinogenicity to huma	the delayed type with onset up to four the delayed type with onset up to four sa sa a group and may not be specific contact eczema, more rarely as urtice hocytes) immune reaction of the delay titions. In the delayed the d	r hours following exposure. c to this product. uria or Quincke's oedema. The pathogenesis of yed type. Other allergic skin reactions, e.g. contact d death. Repeated exposures may alter blood oject of speculation. Fluoroalkanes, in contrast, are mines and ozone react with organic and inorganic veral hundred DBPs have been identified. ation-causing activities. wheezing, gasping, severe distress, even loss of e headache, sleep disturbance, euphoria, r prolonged exposure to irritants may produce Monomers and prepolymers exhibit similar respirator

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	1		
Serious Eye Damage/Irritation	✓	STOT - Single Exposure	✓
Respiratory or Skin sensitisation	✓	STOT - Repeated Exposure	•
Mutagenicity	×	Aspiration Hazard	×

Legend:

💢 – Data either not available or does not fill the criteria for classification 🧪 – Data available to make classification

SECTION 12 Ecological information

Toxicity

Polyset RTA-1- 24oz	Endpoint	Test Duration (hr)	Species	Value	Source
	Not Available	Not Available	Not Available	Not Available	Not Availabl
	Endpoint	Test Duration (hr)	Species	Value	Sourc
	EC50	72h	Algae or other aquatic plants	>100mg/l	2
	EC50	96h	Algae or other aquatic plants	3000-4000mg/l	2
polypropylene glycol	EC50	48h	Crustacea	Crustacea >100mg/l	
	LC50	96h	Fish	>100mg/l	2
	NOEC(ECx)	504h	Crustacea	>=10mg/l	2
	Endpoint	Test Duration (hr)	Species	Value	Sourc
tris(2- chloroisopropyl)phosphate	ErC50	72h	Algae or other aquatic plants	4mg/l	1
	BCF	1008h	Fish	0.8-2.8	7
	EC50(ECx)	96h	Algae or other aquatic plants 4mg/l		1
	EC50	72h	Algae or other aquatic plants 33mg/l		2
	EC50	48h	Crustacea 65335mg/l		1
	LC50	96h	Fish 11mg/l		2
	EC50	96h	Algae or other aquatic plants 4mg/l		1
	Endpoint	Test Duration (hr)	Species	Value	Sourc
	EC50	72h	Algae or other aquatic plants	>170mg/l	2
1,3,3,3-tetrafluoropropene	EC50	48h	Crustacea	>160mg/l	2
	EC50(ECx)	48h	Crustacea	>160mg/l	2
	Endpoint	Test Duration (hr)	Species	Value	Source
	NOEC(ECx)	504h	Crustacea	>=10mg/l	2
4,4'-diphenylmethane diisocyanate (MDI)	BCF	672h	Fish	61-150	7
anocoyanate (mbi)	EC50	72h	Algae or other aquatic plants	>1640mg/l	2
	LC50	96h	Fish	>1000mg/l	2
	Endpoint	Test Duration (hr)	Species	Value	Source
MDI oligomer	Not Available	Not Available	Not Available	Not Available	Not Availab

for polyisocyanates:

Polyisocyanates are not readily biodegradable. However, due to other elimination mechanisms (hydrolysis, adsorption), long retention times in water are not to be expected. The resulting polyurea is more or less inert and, due to its molecular size, not bioavailable.

Data 6. NITE (Japan) - Bioconcentration Data 7. METI (Japan) - Bioconcentration Data 8. Vendor Data

In addition to carbon dioxide (CO2), methane (CH4) and nitrous oxide (N2O), the greenhouse gases mentioned in the Kyoto Protocol include synthetic substances that share the common feature of being highly persistent in the atmosphere and inhibit radiation from escaping out of the atmosphere. These synthetic substances include hydrocarbons that are partially fluorinated (HCFs) or totally fluorinated (PFCs) as well as sulfur hexafluoride (SF6). The greenhouse potential of these substances, expressed as multiples of that of CO2, are within the range of 140 to 11,700 for HFCs, from 6500 to 9,200 for PFCs and 23,900 for SF6. For Isocyanate Monomers:

V3.12 (QSAR) - Aquatic Toxicity Data (Estimated) 4. US EPA, Ecotox database - Aquatic Toxicity Data 5. ECETOC Aquatic Hazard Assessment

Environmental Fate: Isocyanates, (di- and polyfunctional isocyanates), are commonly used to make various polymers, such as polyurethanes. Polyurethanes find significant application in the manufacture of rigid and flexible foams. They are also used in the production of adhesives, elastomers, and coatings.

DO NOT discharge into sewer or waterways

Persistence and degradability

Ingredient	Persistence: Water/Soil	Persistence: Air
tris(2-chloroisopropyl)phosphate	HIGH	HIGH
4,4'-diphenylmethane diisocyanate (MDI)	LOW (Half-life = 1 days)	LOW (Half-life = 0.24 days)

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Ingredient	Bioaccumulation
tris(2-chloroisopropyl)phosphate	LOW (BCF = 4.6)
4,4'-diphenylmethane diisocyanate (MDI)	LOW (BCF = 15)

Mobility in soil

Ingredient	Mobility
tris(2-chloroisopropyl)phosphate	LOW (KOC = 1278)
4,4'-diphenylmethane diisocyanate (MDI)	LOW (KOC = 376200)

SECTION 13 Disposal considerations

Waste treatment methods

- ▶ DO NOT allow wash water from cleaning or process equipment to enter drains.
- Product / Packaging disposal
- It may be necessary to collect all wash water for treatment before disposal.

 In all cases disposal to sewer may be subject to local laws and regulations and these should be considered first.
- ► Consult State Land Waste Management Authority for disposal.
- Discharge contents of damaged aerosol cans at an approved site.
 Allow small quantities to evaporate.

SECTION 14 Transport information

Labels Required



Marine Pollutant

Land transport (DOT)

UN number	1950		
UN proper shipping name	Aerosols, non-flammable, (each not exceeding 1 L capacity)		
Transport hazard class(es)	Class 2.2 Subrisk Not Applicable		
Packing group	Not Applicable		
Environmental hazard	Not Applicable		
Special precautions for user	Hazard Label 2.2 Special provisions Not Applicable		

Air transport (ICAO-IATA / DGR)

UN number	1950			
UN proper shipping name	Aerosols, non-flammable; Aerosols, non-flammable (containing biological products or a medicinal preparation which will be deteriorated by a heat test)			
Transport hazard class(es)	ICAO/IATA Class 2.2 ICAO / IATA Subrisk Not Applicable ERG Code 2L			
Packing group	Not Applicable			
Environmental hazard	Not Applicable			
Special precautions for user	Special provisions Cargo Only Packing Instructions Cargo Only Maximum Qty / Pack Passenger and Cargo Packing Instructions Passenger and Cargo Maximum Qty / Pack Passenger and Cargo Limited Quantity Packing Instructions Passenger and Cargo Limited Maximum Qty / Pack		A98 A145 A167 A802 203 150 kg 203 75 kg Y203 30 kg G	

Sea transport (IMDG-Code / GGVSee)

UN number	195
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IIN was a shinning name	AEBOSOLO	
UN proper shipping name	AEROSOLS	
Transport hazard class(es)	IMDG Class 2	2.2
	IMDG Subrisk 1	Not Applicable
Packing group	Not Applicable	
Environmental hazard	Not Applicable	
	EMS Number	F-D , S-U
Special precautions for user	Special provisions	63 190 277 327 344 381 959
	Limited Quantities	1000 ml

Transport in bulk according to Annex II of MARPOL and the IBC code

Not Applicable

Transport in bulk in accordance with MARPOL Annex V and the IMSBC Code

Product name	Group
polypropylene glycol	Not Available
tris(2-chloroisopropyl)phosphate	Not Available
1,3,3,3-tetrafluoropropene	Not Available
4,4'-diphenylmethane diisocyanate (MDI)	Not Available
MDI oligomer	Not Available

Transport in bulk in accordance with the ICG Code

Product name	Ship Type
polypropylene glycol	Not Available
tris(2-chloroisopropyl)phosphate	Not Available
1,3,3,3-tetrafluoropropene	Not Available
4,4'-diphenylmethane diisocyanate (MDI)	Not Available
MDI oligomer	Not Available

SECTION 15 Regulatory information

Safety, health and environmental regulations / legislation specific for the substance or mixture

polypropylene glycol is found on the following regulatory lists

US DOE Temporary Emergency Exposure Limits (TEELs) US Toxic Substances Control Act (TSCA) - Chemical Substance Inventory

US Toxicology Excellence for Risk Assessment (TERA) Workplace Environmental Exposure Levels (WEEL)

US TSCA Chemical Substance Inventory - Interim List of Active Substances

tris(2-chloroisopropyl)phosphate is found on the following regulatory lists

US Toxic Substances Control Act (TSCA) - Chemical Substance Inventory

1,3,3,3-tetrafluoropropene is found on the following regulatory lists

US DOE Temporary Emergency Exposure Limits (TEELs)

US Toxic Substances Control Act (TSCA) - Chemical Substance Inventory

US TSCA Chemical Substance Inventory - Interim List of Active Substances

US TSCA Chemical Substance Inventory - Interim List of Active Substances

US Toxicology Excellence for Risk Assessment (TERA) Workplace Environmental

4,4'-diphenylmethane diisocyanate (MDI) is found on the following regulatory lists

International Agency for Research on Cancer (IARC) - Agents Classified by the IARC Monographs

US - California Hazardous Air Pollutants Identified as Toxic Air Contaminants

US ACGIH Threshold Limit Values (TLV)

US AIHA Workplace Environmental Exposure Levels (WEELs)

US Clean Air Act - Hazardous Air Pollutants

US DOE Temporary Emergency Exposure Limits (TEELs)

US EPA Integrated Risk Information System (IRIS)

US EPCRA Section 313 Chemical List

Exposure Levels (WEEL)

US NIOSH Recommended Exposure Limits (RELs)

US OSHA Permissible Exposure Limits (PELs) Table Z-1

US Toxic Substances Control Act (TSCA) - Chemical Substance Inventory

US TSCA Chemical Substance Inventory - Interim List of Active Substances

US TSCA New Chemical Exposure Limits (NCEL)

MDI oligomer is found on the following regulatory lists

International Agency for Research on Cancer (IARC) - Agents Classified by the IARC Monographs

US ATSDR Minimal Risk Levels for Hazardous Substances (MRLs)

US DOE Temporary Emergency Exposure Limits (TEELs)

US EPCRA Section 313 Chemical List

US Toxic Substances Control Act (TSCA) - Chemical Substance Inventory

US TSCA Chemical Substance Inventory - Interim List of Active Substances

Federal Regulations

Superfund Amendments and Reauthorization Act of 1986 (SARA)

Section 311/312 hazard categories

Flammable (Gases, Aerosols, Liquids, or Solids)	
Gas under pressure	Yes

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Explosive	No
Self-heating	No
Pyrophoric (Liquid or Solid)	No
Pyrophoric Gas	No
Corrosive to metal	No
Oxidizer (Liquid, Solid or Gas)	No
Organic Peroxide	No
Self-reactive	No
In contact with water emits flammable gas	No
Combustible Dust	No
Carcinogenicity	No
Acute toxicity (any route of exposure)	Yes
Reproductive toxicity	No
Skin Corrosion or Irritation	Yes
Respiratory or Skin Sensitization	Yes
Serious eye damage or eye irritation	Yes
Specific target organ toxicity (single or repeated exposure)	Yes
Aspiration Hazard	No
Germ cell mutagenicity	No
Simple Asphyxiant	No
Hazards Not Otherwise Classified	No

US. EPA CERCLA Hazardous Substances and Reportable Quantities (40 CFR 302.4)

Name	Reportable Quantity in Pounds (lb)	Reportable Quantity in kg
4,4'-diphenylmethane diisocyanate (MDI)	5000	2270

State Regulations

US. California Proposition 65

None Reported

National Inventory Status

National Inventory	Status		
Australia - AIIC / Australia Non-Industrial Use	Yes		
Canada - DSL	Yes		
Canada - NDSL	No (polypropylene glycol; tris(2-chloroisopropyl)phosphate; 4,4'-diphenylmethane diisocyanate (MDI); MDI oligomer)		
China - IECSC	No (1,3,3,3-tetrafluoropropene)		
Europe - EINEC / ELINCS / NLP	Yes		
Japan - ENCS	Yes		
Korea - KECI	Yes		
New Zealand - NZIoC	No (1,3,3,3-tetrafluoropropene)		
Philippines - PICCS	No (1,3,3,3-tetrafluoropropene)		
USA - TSCA	Yes		
Taiwan - TCSI	Yes		
Mexico - INSQ	No (1,3,3,3-tetrafluoropropene)		
Vietnam - NCI	Yes		
Russia - FBEPH	Yes		
Legend:	Yes = All CAS declared ingredients are on the inventory No = One or more of the CAS listed ingredients are not on the inventory and are not exempt from listing(see specific ingredients in brackets)		

SECTION 16 Other information

Revision Date	04/28/2021
Initial Date	04/20/2021

CONTACT POINT

PLEASE NOTE THAT TITANIUM DIOXIDE IS NOT PRESENT IN CLEAR OR NEUTRAL BASES

SDS Version Summary

Version	Date of Update	Sections Updated
1.7.2.1	04/28/2021	Name

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Other information

Classification of the preparation and its individual components has drawn on official and authoritative sources as well as independent review by the Chemwatch Classification committee using available literature references.

The SDS is a Hazard Communication tool and should be used to assist in the Risk Assessment. Many factors determine whether the reported Hazards are Risks in the workplace or other settings. Risks may be determined by reference to Exposures Scenarios.

Definitions and abbreviations

PC-TWA: Permissible Concentration-Time Weighted Average PC-STEL: Permissible Concentration-Short Term Exposure Limit

IARC: International Agency for Research on Cancer

ACGIH: American Conference of Governmental Industrial Hygienists

STEL: Short Term Exposure Limit

TEEL: Temporary Emergency Exposure Limit。

IDLH: Immediately Dangerous to Life or Health Concentrations

ES: Exposure Standard OSF: Odour Safety Factor

NOAEL :No Observed Adverse Effect Level

LOAEL: Lowest Observed Adverse Effect Level

TLV: Threshold Limit Value LOD: Limit Of Detection

OTV: Odour Threshold Value BCF: BioConcentration Factors

BEI: Biological Exposure Index

AIIC: Australian Inventory of Industrial Chemicals

DSL: Domestic Substances List

NDSL: Non-Domestic Substances List

IECSC: Inventory of Existing Chemical Substance in China

EINECS: European INventory of Existing Commercial chemical Substances

ELINCS: European List of Notified Chemical Substances

NLP: No-Longer Polymers

ENCS: Existing and New Chemical Substances Inventory

KECI: Korea Existing Chemicals Inventory NZIoC: New Zealand Inventory of Chemicals

PICCS: Philippine Inventory of Chemicals and Chemical Substances

TSCA: Toxic Substances Control Act

TCSI: Taiwan Chemical Substance Inventory

INSQ: Inventario Nacional de Sustancias Químicas

NCI: National Chemical Inventory

FBEPH: Russian Register of Potentially Hazardous Chemical and Biological Substances

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